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Raoult's law. A law stating that the partial pressure of a solvent over a solution, P_{solution} , is given by the vapor pressure of the pure solvent, $P^{\circ}_{\text{solvent}}$, times the mole fraction of a solvent in the solution, X_{solvent} : $P_{\text{solution}} = X_{\text{solvent}}P^{\circ}_{\text{solvent}}$. (Section 13.5)

Regioselective reaction.

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