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Stoichiometry | ShowMe

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08-25T00:00:00+00:01 Subject: Skill
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Skill Practice 30 #'s 1-5

Skill Practice 1 Name: _____ Date: _____
Hour: _____ For problems 1-3, please use
these conversion factors: 1 pallet = 45
bundles 1 bundle = 32 cases 1 case =
12 cans 1 can = 218.4 mL 1. How many
mL of apple juice does the company
need to make to fulfill an order for 2.5
pallets? 9,434,880 mL 2. If 46,680 mL of

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juice are produced, how many ...

Skill Practice 1

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Partial Pressure, Effusion - Duration:
2:00:12. The Organic Chemistry Tutor
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Skill Practice 31

How many moles of Iron (Fe) are present in a sample containing 4.41×10^{22} atoms? moles
How many molecules of Oxygen (O_2) are present in a sample that is 1.83×10^{-23} moles? molecules
How many moles of Hydrogen (H_2) are present in a sample weighs 8.57 g? moles

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Mass and Mole Relations Exercises

over and over again, chemists refer to it as a mole (abbreviated mol). 6.022 10²³ objects is called a mole, just as you call 12 objects a dozen. Look again at how these quantities are related. 55.847 g of iron 6.022 10²³ iron atoms 1 mol of iron 32.066 g of sulfur 6.022 10²³ sulfur

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atoms 1 mol of sulfur General Plan for
Converting Mass, Amount,

Skills Worksheet Problem Solving

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Skill Practice 30 Conversion Practice

Name: Date: Hour: I. Calculate the number of molecules in 210 grams of

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'water. 7.02×10^2 molecules 2. If you have 6.25×10^2 molecules of sulfur tetrafluoride, how many grams do you have?

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Skill Practice 36 Name: _____ Date: _____

Hour: _____ 1. Under water where the temperature is 17°C and the pressure is 394 kPa, a diver inhales 2.1 L of air from

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his SCUBA tank. a) How many moles of gas are in his lungs? 0.343 mol b) If the diver swims to the surface without exhaling where the temperature is 32oC and the

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