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Answers may vary. Sample answer: The collision theory states that a chemical reaction occurs when reactants collide in the right orientation and with enough energy to break chemical bonds and make new ones. For example, in (a) and (b) below, the orientations of the bromine atoms allow them to contact one another, so the chemical reaction can occur.

Chapter 6 Review, pages 396-401

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The video below refers to Dynamic Equilibrium (under the section Equilibrium Systems) 7.1 p. 420 in the Nelson Textbook The video below is related to 7.2 Equilibrium Law and the Equilibrium Constant (p.429 in the Textbook)

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