

Chapter 9 Stoichiometry Review Answers

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Q. 4.3 grams of sodium reacts with 2.6 grams of oxygen to produce 5.9 grams of sodium oxide. What's the limiting reactant?

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Modern Chemistry 77 Stoichiometry CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. _____ The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N_2 are mixed with 12.0 mol of H_2

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Stoichiometry SHORT ANSWER Answer the following questions in the space provided. 1. b The coefficients in a chemical equation represent the (a) masses in grams of all reactants and products. (h) relative number of moles of reactants and products. (c) number of atoms of each element in each compound a reaction.

Date. FCHAPJ REV[EW].

Stoichiometry problems can be characterized by two things: (1) the information given in the problem, and (2) the information that is to be solved for, referred to as the unknown. The given and the unknown may both be reactants, both be products, or one may be a reactant while the other is a product.

Stoichiometry | Chemistry for Non-Majors

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CHAPTER 9 REVIEW. Stoichiometry. SECTION 9.2. PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. The following equation represents a laboratory preparation for oxygen gas:

CHAPTER 9 REVIEW

CHAPTER 9 REVIEW. Stoichiometry. SECTION 1. SHORT ANSWER Answer the following questions in the space provided. 1. ____ The coefficients in a chemical equation represent the. (a) masses in grams of all reactants and products. (b) relative number of moles of reactants and products. (c) number of atoms of each element in each compound in a reaction.

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